## Contents

Preface			xiii	
Acknowledgments				liii
1.	1. INTRODUCTION			1
	1 Axiomatic Consistency of Quantum Chemistry			1
	2	Scope of	of These Studies	2
	3	Insuffic Structu	iencies of Quantum Chemistry for Molecular tres	5
	4	Insuffic Reactio	iency of Quantum Chemistry for Chemical ons	11
	5	Insuffic Structu	iencies of Quantum Chemistry for Biological res	12
6 The Central Topic of Study of This Monograp			entral Topic of Study of This Monograph	14
	7		rophic Inconsistencies of Generalized Nonunitary es on Conventional Mathematics	19
	8	Hadron	nic Mechanics	27
	9	Hadron	nic Superconductivity	34
2.			S OF ISO-, GENO-, AND HYPER-MATHEMATICS R ISODUALS	59
	1	Introdu	uction	59
	2 Elements of Isomathematics		nts of Isomathematics	61
		2.1	Isounits and Isoproducts	61
		2.2	Isonumbers and Isofields	64
		2.3	Isospaces and Isogeometries	65
		2.4	Isodifferential Calculus	69
		2.5	Isohilbert Spaces	70
		2.6	Isoperturbation Theory	72
		2.7	Isofunctional Analysis	74

## Contents

4.

5.

		2.8	Isolinearity, Isolocality, Isocanonicity and	
			Isounitarity	75
		2.9	Lie-Santilli Isotheory	77
	3	Elemen	nts of Genomathematics	81
		3.1	Introduction	81
		3.2	Main Structural Lines of Genomathematics	82
		3.3	Genounits and Genoproducts	88
		3.4	Genonumbers and Genofields	89
		3.5	Genospaces and Genogeometries	91
		3.6	Genodifferential Calculus	92
		3.7	Genohilbert Spaces	93
		3.8	Genolinearity, Genolocality, and Genounitarity	94
		3.9	Lie-Santilli Genotheory	95
	4		mathematics	97
	5	Isodua	l Mathematics	99
3.	FOU	UNDAT	TIONS OF HADRONIC CHEMISTRY	107
	1	Introd	uction	107
	2	Classic	eal Foundations of Hadronic Chemistry	111
		2.1	The Historical Teaching of Lagrange and	
			Hamilton	111
		2.2	The Inevitability of the Historical Teaching	112
		2.3	Problematic Aspects of External Terms	113
		2.4	Classification of Hamilton's Equation with External	
			Terms	114
		2.5	Hamilton-Santilli Isomechanics	115
		2.6	Classical Lie-Santilli Brackets	118
		2.7	Isoaction Principle	120
		2.8	Hamilton-Jacobi-Santilli Isoequations	121
		2.9	Examples of Classical Applications	121
		2.10	Connection Between Isotopic and Birkhoffian	
			Mechanics	122
		2.11	Hamilton-Santilli Geno-, Hyper-, and Isodual-	
			Mechanics	124
		2.12	Simple Construction of Classical Isochemistry	126
		2.13	Invariance of Classical Isochemistry	128
		2.14	Simple Construction of Classical Genochemistry	129
		2.15	Invariance of Classical Genochemistry	129
		2.16	Simple Construction and Invariance of Hyper- and Isodual Mechanics	129
	3	Operat	tor Foundations of Hadronic Chemistry	130
		3.1	Introduction	130
		3.2	Naive Iso-, Geno, Hyper-, and Isodual	
			Quantization	130
		3.3	Structure of Operator Isochemistry	132
			~ v	

	3.4	Basic Equations of Operator Isochemistry	134
	3.5	Preservation of Quantum Physical Laws	135
	3.6	Simple Construction of Operator Isochemistry	139
	3.7	Invariance of Operator Isochemistry	142
	3.8	Gaussian Screenings as Particular Cases of	
		Isochemistry	143
	3.9	Elements of Operator Geno-, Hyper-, and Isodual-	
	0.10	Chemistry	144
	3.10	Simple Construction of Operator Geno-, Hyper-,	
	9.11	and Isodual Chemistry	147
	3.11	Invariance of Operator Geno-, Hyper-, and Isodual	140
	3.12	Chemistry Classification of Hadronic Chemistry	$\frac{149}{149}$
	3.12	Classification of Hadronic Chemistry	149
ISO	CHEM	ICAL MODEL OF THE HYDROGEN MOLECULE	2155
1	Introd	uction	155
2	Isochei	mical Model of Molecular Bonds	158
3	The Li	mit Case of Stable Isoelectronium	168
4		mical Model of the Hydrogen Molecule with Stable	
Т		tronium	174
5		y Solvable, Three-Body, Isochemical Model of the	1.1
0		gen Molecule	177
6			<b>T</b> 1 1
0		mical Model of the Hydrogen Molecule with Unstable tronium	180
7			100
1		an Approximation of the Isochemical Model of the gen Molecule as a Four-Body System	183
8		ary of the Results	188
9	-		
Appendix 4.A Isochemical Calculations for the Three-Body $H_2$ Molecule			195
App		.B Isochemical Calculations for the Four-Body $H_2$	100
трр	Moleci		196
	MOICCI		150
ISO	CHEM	ICAL MODEL OF THE WATER MOLECULE	205
1	Introd	uction	205
2	Main (	Characteristics of Water	211
3	Exactl	y Solvable Model of the Water Molecule with Stable	
		tronium	215
4	Gaussi	an Approximation of the Isochemical Model of the	
		Molecule with Unstable Isoelectronium	219
5	The M	ethod	223
6		ain Results	226
7	Conclu		228
1	Concit	1910119	220

ix

## Contents

6.	VARIATIONAL CALCULATIONS OF ISOCHEMICAL MOLECULAR MODELS			
	1	Introduction	233 233	
	2	Aringazin-Kucherenko Study of the Restricted, Three-Body Isochemical Model of the Hydrogen Molecule	234	
	3	Aringazin Variational Study of the Four-Body Isochemical Model of the Hydrogen Molecule	239	
7.	APPLICATION OF HADRONIC CHEMISTRY TO NEW CLEAN ENERGIES AND FUELS			
	1	Introduction	251	
	2	Alarming Environmental Problems Caused by Gasoline and Coal Combustion	254	
	3	Alarming Environmental Problems Caused by Natural Gas Combustion	257	
	4	Alarming Environmental Problems Caused by Hydrogen Combustion, Fuel Cells and Electric Cars	258	
	5	The Need for New, Environmentally Acceptable Primary Sources of Electricity	261	
	6	Insufficiencies of Quantum Mechanics, Superconductivity, and Chemistry for the Solution of Current Environmental Problems	262	
	7	The New Clean Primary Energies Predicted by Hadronic Mechanics, Superconductivity and Chemistry	263	
	8	PlasmaArcFlow Reactor for the Conversion of Liquid Waste into the Clean Burning Magnegas	279	
	9	Surpassing by Magnegas Exhaust of EPA Requirements without Catalytic Converter	283	
	10	Anomalous Chemical Composition of Magnegas	288	
	11	Anomalous Energy Balance of Hadronic Molecular Reactors	291	
	12	Concluding Remarks	298	
	Appendix 7.A		300	
8.	TH	E NEW CHEMICAL SPECIES OF MAGNECULES	303	
	1	Introduction	303	
	2	The Hypothesis of Magnecules	305	
	3	The Five Force Fields Existing in Atoms	308	
	4	Magnecules Internal Bonds	310	
	5	Production of Magnecules in Gases, Liquids and Solids	316	
	6	New Molecules Internal Bonds	322	
	7	Main Features of Magnecules to be Detected	324	

8	Necessary Conditions for the Correct Detection of Molecules		
	and N	Magnecules	326
	8.1	Selection of Analytic Instruments	326
	8.2	Unambiguous Detection of Molecules	328
	8.3	Unambiguous Detection of Magnecules	331
	8.4	Apparent Magnecular Structure of $H_3$ and $O_3$	333
	8.5	Need for New Analytic Methods	336
9	Experimental Evidence of Gas Magnecules		337
	$9.1^{-1}$	Conventional Chemical Composition of Magnegas	
		Used in the Tests	337
	9.2	GC-MS/IRD Measurements of Magnegas at the	
		McClellan Air Force Base	338
	9.3	GC-MS/IRD Tests of Magnegas at Pinellas County	
		Forensic Laboratory	341
	9.4	Interpretations of the Results	347
	9.5	Concluding Remarks	352
10	Experimental Evidence of Liquid Magnecules		354
	10.1	Preparation of Liquid Magnecules used in the	
		Tests	354
	10.2	Photographic Evidence of Magnecules in Liquids	355
	10.3	Spectroscopic Evidence of Liquid Magnecules at	
		the Tekmar-Dohrmann Corporation	358
	10.4	Spectroscopic Evidence of Liquid Magnecules at	
		Florida International University	363
11	Experimental Verification of Mutated Physical		
	Characteristics		364
12	Concl	luding Remarks	375
App	opendix 8.A Aringazin's Studies on Toroidal Orbits of the		
11		ogen Atom under an External Magnetic Field	376
T 1	•		202
Index			393